

1 15A NCAC 11 .0303 is proposed for amendment as follows:

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3 **15A NCAC 11 .0303 EXEMPT CONCENTRATIONS: OTHER THAN SOURCE MATERIAL**

4 (a) No person shall introduce radioactive material into a product or material knowing or having reason to believe
5 that it will be transferred to persons exempt under Paragraph ~~(b)~~ (d) of this Rule or equivalent regulations of the U.S.
6 Nuclear Regulatory Commission or any agreement state, except in accordance with a specific license issued
7 pursuant to ~~Rule .0325 of this Section~~ 10 CFR 32.11.

8 (b) A manufacturer, processor, or producer of a product or material is exempt from the requirements for a license
9 set forth in these rules to the extent that this person transfers radioactive material contained in a product or material
10 in concentrations not in excess of those specified in paragraph (d) of this rule, and introduced into the product or
11 material by a licensee holding a specific license issued by the US Nuclear Regulatory Commission expressly
12 authorizing such introduction. This exemption does not apply to the transfer of byproduct material contained in any
13 food, beverage, cosmetic, drug, or other commodity designed for ingestion or inhalation by, or application to, a
14 human being.

15 (c) This rule shall not be deemed to authorize the import of radioactive material or products containing radioactive
16 material.

17 ~~(b)~~ (d) Except as provided in Paragraph (a) and (b) of this Rule, any person is exempt from these Rules to the extent
18 that such person receives, possesses, uses, transfers, owns, or acquires products or materials containing radioactive
19 material in concentrations not in excess of those listed in the following table:

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21 EXEMPT CONCENTRATIONS

Element <u>(atomic number)</u>	Isotope	Column I concentration microcurie/ml	Column II Liquid and solid concentration microcurie/ml
Antimony (51)	Sb 122		3×10^4
	Sb 124		2×10^4
	Sb 125		1×10^3
Argon (18)	Ar 37	1×10^3	
	Ar 41	4×10^{-7}	
Arsenic (33)	As 73		5×10^{-3}
	As 74		5×10^{-4}
	As 76		2×10^{-4}
	As 77		8×10^{-4}
Barium (56)	Ba 131		2×10^{-3}

1		Ba 140		3X10 ⁴
2	Beryllium (4)	Be 7		2X10 ²
3	Bismuth (83)	Bi 206		4X10 ⁴
4	Bromine (35)	Br 82	4X10 ⁷	3X10 ³
5	Cadmium (48)	Cd 109		2X10 ³
6		Cd 115m		3X10 ⁴
7		Cd 115		3X10 ⁴
8	Calcium (20)	Ca 45		9X10 ⁵
9		Ca 47		5X10 ⁴
10	Carbon (6)	C 14	1X10 ⁶	8X10 ³
11	Cerium (58)	Ce 141		9X10 ⁴
12		Ce 143		4X10 ⁴
13		Ce 144		1X10 ⁴
14	Cesium (55)	Cs 131		2X10 ²
15		Cs 134m		6X10 ²
16		Cs 134		9X10 ⁵
17	Chlorine (17)	Cl 38	9X10 ⁷	4X10 ³
18	Chromium (24)	Cr 51		2X10 ²
19	Cobalt (27)	Co 57		5X10 ³
20		Co 58		1X10 ³
21		Co 60		5X10 ⁴
22	Copper (29)	Cu 64		3X10 ³
23	Dysprosium (66)	Dy 165		4X10 ³
24		Dy 166		4X10 ⁴
25	Erbium (68)	Er 169		9X10 ⁴
26		Er 171		1X10 ³
27	Europium (63)	Eu 152		6X10 ⁴
28		(T1/2 =9.2 Hrs.)		
29		Eu 155		2X10 ³
30	Fluorine (9)	F 18	2X10 ⁶	8X10 ³
31	Gadolinium (64)	Gd 153		2X10 ³
32		Gd 159		8X10 ⁴
33	Gallium (31)	Ga 72		4X10 ⁴
34	Germanium (32)	Ge 71		2X10 ²
35	Gold (79)	Au 196		2X10 ³
36		Au 198		5X10 ⁴

1		Au 199	2X10 ⁻³
2	Hafnium (72)	Hf 181	7X10 ⁻⁴
3	Hydrogen (1)	H 3	3X10 ⁻²
4	Indium (49)	In 113m	1X10 ⁻²
5		In 114m	2X10 ⁻⁴
6	Iodine (53)	I 126	2X10 ⁻⁵
7		I 131	2X10 ⁻⁵
8		I 132	6X10 ⁻⁴
9		I 133	7X10 ⁻⁵
10		I 134	1X10 ⁻³
11	Iridium (77)	Ir 190	2X10 ⁻³
12		Ir 192	4X10 ⁻⁴
13		Ir 194	3X10 ⁻⁴
14	Iron (26)	Fe 55	8X10 ⁻³
15		Fe 59	6X10 ⁻⁴
16	Krypton (36)	Kr 85m	<u>1X10⁻⁶</u>
17		Kr 85	<u>3X10⁻⁶</u>
18	Lanthanum (57)	La 140	2X10 ⁻⁴
19	Lead (82)	Pb 203	4X10 ⁻³
20	Lutetium (71)	Lu 177	1X10 ⁻³
21	Manganese (25)	Mn 52	3X10 ⁻⁴
22		Mn 54	1X10 ⁻³
23		Mn 56	1X10 ⁻³
24	Mercury (80)	Hg 197m	2X10 ⁻³
25		Hg 197	3X10 ⁻³
26		Hg 203	2X10 ⁻⁴
27	Molybdenum (42)	Mo 99	2X10 ⁻³
28	Neodymium (60)	Nd 147	<u>6X10⁻³</u> <u>6X10⁻⁴</u>
29		Nd 149	<u>3X10⁻⁴</u> <u>3X10⁻³</u>
30	Nickel (28)	Ni 65	1X10 ⁻³
31	Niobium(Columbium)(41)	Nb 95	1X10 ⁻³
32		Nb 97	9X10 ⁻³
33	Osmium (76)	Os 185	7X10 ⁻⁴
34		Os 191m	3X10 ⁻²
35		Os 191	2X10 ⁻³
36		Os 193	6X10 ⁻⁴

1	Palladium (46)	Pd 103	3×10^{-3}
2		Pd 109	9×10^{-4}
3	Phosphorus (15)	P 32	2×10^{-4}
4	Platinum (78)	Pt 191	1×10^{-3}
5		Pt 193m	1×10^{-2}
6		Pt 197m	1×10^{-2}
7		Pt 197	1×10^{-3}
8	<u>Polonium (84)</u>	Po 210	<u>7×10^{-6}</u>
9	Potassium (19)	K 42	3×10^{-3}
10	Praseodymium (59)	Pr 142	3×10^{-4}
11		Pr 143	5×10^{-4}
12	Promethium (61)	Pm 147	2×10^{-3}
13		Pm 149	4×10^{-4}
14	<u>Radium (88)</u>	Ra 226	<u>1×10^{-7}</u>
15		Ra 228	<u>3×10^{-7}</u>
16	Rhenium (75)	Re 183	6×10^{-3}
17		Re 186	9×10^{-4}
18		Re 188	6×10^{-4}
19	Rhodium (45)	Rh 103m	1×10^{-1}
20		Rh 105	1×10^{-3}
21	Rubidium (37)	Rb 86	7×10^{-4}
22	Ruthenium (44)	Ru 97	4×10^{-3} <u>4×10^{-4}</u>
23		Ru 103	8×10^{-4}
24		Ru 105	1×10^{-3}
25		Ru 106	1×10^{-4}
26	Samarium (62)	Sm 153	8×10^{-4}
27	Scandium (21)	Sc 46	4×10^{-4}
28		Sc 47	9×10^{-4}
29		Sc 48	3×10^{-4}
30	Selenium (34)	Se 75	3×10^{-3}
31	Silicon (14)	Si 31	9×10^{-3}
32	Silver (47)	Ag 105	1×10^{-3}
33		Ag 110m	3×10^{-4}
34		Ag 111	4×10^{-4}
35	Sodium (11)	Na 24	2×10^{-3}
36	Strontium (38)	Sr 85	<u>4×10^{-3}</u> <u>1×10^{-4}</u>

1		Sr 89		1X10 ⁴
2		Sr 91		7X10 ⁴
3		Sr 92		7X10 ⁴
4	Sulfur (16)	S 35	9X10 ⁸	6X10 ⁴
5	Tantalum (73)	Ta 182		4X10 ⁴
6	Technetium (43)	Tc 96m		1X10 ¹
7		Tc 96		1X10 ³
8	Tellurium (52)	Te 125m		2X10 ³
9		Te 127m		6X10 ⁴
10		Te 127		3X10 ³
11		Te 129m		3X10 ⁴
12		Te 131m		6X10 ⁴
13		Te 132		3X10 ⁴
14	Terbium (65)	Tb 160		4X10 ⁴
15	Thallium (81)	Tl 200		4X10 ³
16		Tl 201		3X10 ³
17		Tl 202		1X10 ³
18		Tl 204		1X10 ³
19	Thulium (69)	Tm 170		5X10 ⁴
20		Tm 171		5X10 ³
21	Tin (50)	Sn 113		9X10 ⁴
22		Sn 125		2X10 ⁴
23	Tungsten(Wolfram) (74)	W 181		4X10 ³
24		W 187		7X10 ⁴
25	Vanadium (23)	V 48		3X10 ⁴
26	Xenon (54)	Xe 131m		4X10 ⁶
27		Xe 133		3X10 ⁶
28		Xe 135		1X10 ⁶
29	Ytterbium (70)	Yb 175		1X10 ³
30	Yttrium (39)	Y 90		2X10 ⁴
31		Y 91m		3X10 ²
32		Y 91		3X10 ⁴
33		Y 92		6X10 ⁴
34		Y 93		3X10 ⁴
35	Zinc (30)	Zn 65		1X10 ³
36		Zn 69m		7X10 ⁴

1	Zn 69	2X10 ²
2	Zirconium (40)	6X10 ⁴
3	Zr 95	2X10 ⁴
4	Beta and/or gamma emitting radioactive material not listed above with half-life less than 3 years	1X10 ⁻¹⁰
5		1X10 ⁻⁶
6		
7		
8		

9 ~~(e)~~ (e) In Column I of the table, in Paragraph (b) of this Rule, values are given only for those materials normally
10 used as gases.

11 ~~(f)~~ (f) In Column II of the table, in Paragraph (b) of this Rule, the units, microcuries per gram, are used for solids.

12 ~~(g)~~ (g) Many radioisotopes disintegrate into isotopes which are also radioactive. In expressing the concentrations in
13 Paragraph (b) of this Rule, the activity stated is that of the parent isotope and takes into account the daughters.

14 ~~(h)~~ (h) For purposes of this Rule, where a combination of isotopes is involved, the limit for the combination shall be
15 derived as follows: Determine for each isotope in the product the ratio between the concentration present in the
16 product and the exempt concentration established in Paragraph (b) of this Rule for the specific isotope when not in
17 combination. The sum of the ratios shall not exceed unity. An example of this is:

18
19 Concentration of Isotope A in Product +
20 Exempt concentration of Isotope A

21
22 Concentration of Isotope B in Product less than or equal to 1
23 Exempt concentration of Isotope B

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25
26 *History Note:* *Authority G.S. 104E-7; 104E-10; 104E-20;*

27 *Eff. February 1, 1980;*

28 *Amended Eff. October 1, 2013; May 1, 1993; June 1, 1989.*